# Evaporation of Liquid Chlorides in Closed Tanks using the PHOENICS Code

Presentation of a Rigid Interface Model (RIModel)

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#### **SUMMARY**

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- 5 CONCLUSION

#### **INTRODUCTION**

- 1 Optical Fiber Industry
- Liquid (SiCl<sub>4</sub>, GeCl<sub>4</sub>, ...) changed into Vapor and Oxydized (SiO<sub>2</sub>, GeO<sub>2</sub>, ...)
- Mainly used Current Technique : Bubbling ⇒ Carrier Gas send into Liquid
- Intended Technique : Direct Liquid Evaporation for Higher Vapor Flow expected
- 2 Prediction of Evaporation Phenomenon means turn to Numerical Simulation
- Hydrodynamic and Thermal Phenomena Simulation inside Closed Tanks
- Start from Equilibrium Situation and Simulation of Flow Requirements for Optical Fiber Production



#### INTRODUCTION

- 3 Specific Model for prediction of Evaporation Process in closed Tanks
- Multi-Domain Method
  - ⇒ Resolution of each Phase separately
    - ⇒ Two Phases linked by Boundaries Conditions at the Interface
- Use CHAM's CFD PHOENICS
  - ⇒ Computation of a Specific Model for Evaporation in Closed Tanks
  - ⇒ Interface taken as a Rigid Plate with a Moving Speed
    - ⇒ RIModel (acronym of Rigid Interface Model)



## **EVAPORATION OF LIQUIDS - Theorical Background**

- 1 Two kinds of Evaporation Processes for Liquids
- Vapor directly produced at Liquid / Vapor Interface (Radiative Heating for instance)
- Vapor produced by Bubbles starting at Immersed Solid Heated Surface (Pool Boiling)
- <u>2 Pool Boiling</u>  $\Rightarrow$  4 Boiling Mechanisms
- Subcooled Boiling  $\Rightarrow$  No bubbles (or Recondense near Heated Surface)
- Boiling with Net Evaporation  $\Rightarrow$  *Nucleate Boiling Region* 
  - ⇒ Partial Film Boiling Region (or Transition)
  - $\Rightarrow$  Film Boiling Region



## **EVAPORATION OF LIQUIDS - Theorical Background**

3 - How to predict Evaporation Phenomenon in Closed Tanks?

#### $\Rightarrow$ ASSUMPTION :

*Pool Subcooled Boiling* ⇒ *Pressure Stability, Avoiding Decay of Liquid Precursors* 

#### ⇒ CONSEQUENCE :

Two Phases Liquid and Vapor strictly separated: Analogy with Melting Process

- ⇒ Two Regions with Different Thermodynamic Properties separated by a Moving Interface
- ⇒ Heat Exchange at the Interface according Heat Conduction with Release (Solidification or Condensation Process) or Absorption of Heat (Melting or Evaporation Process)

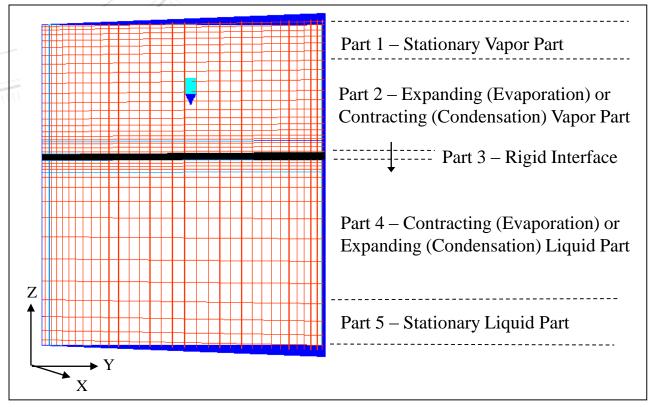
## RIModel - Governing Equations and Characteristics

- 4 Implement Features with PHOENICS ⇒ Rigid Interface Model (RIModel)
  - Vapor Quantity Calculation by a Mass Balance and the Perfect Gas Law: Mvap (t)
     Mvap (t) = Mvap(t = 0) Mcoll (t) + Mint (t) ⇒ Pvap
  - Liquid Interface Temperature Calculation with Antoine's Equation (SiCl4): **Ts**  $Pvap = 10^{4}.09777 1200 / (Ts 37) \Rightarrow Ts$
  - Mass Exchange Calculation at the Interface : **qint** (**t**) qint (t) .  $\Delta Hvap = [\lambda vap . (\partial T/\partial z) \lambda liq . (\partial T/\partial z)] \Rightarrow$ **qint**(**t**)
  - Interface taken as Rigid Plate with Moving Speed equal to Evaporation Speed
  - Friction at the Vapor Side of the Plate and Slip without Friction at the Liquid Side of the Plate



#### RIModel - Application with the PHOENICS Code

- 1 Moving Grid Option ⇒ ZMOVE function of PHOENICS
- 2 Tank Internal Domain splitted in 5 Zones



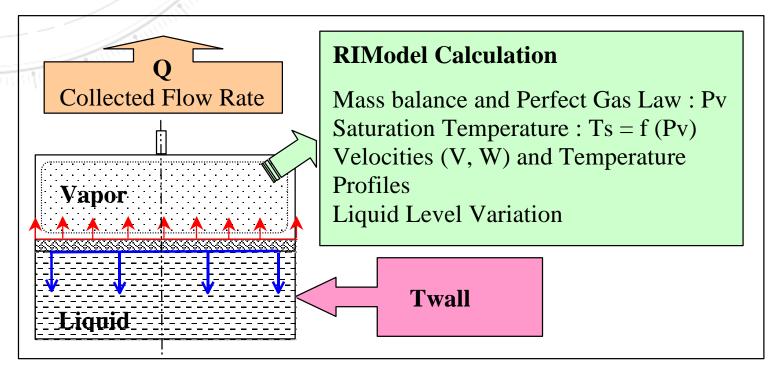
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## RIModel - Boundaries Conditions for Parametric Study

3 PARAMETERS FOR
PARAMETRIC STUDY

- $\Rightarrow$  Q ( Collected Vapor Flow )
- $\Rightarrow$  S ( Size of Evaporation Surface )
- $\Rightarrow$  Twall ( Fixed Wall Temperature )



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#### RIModel - Simulated Cases

		S1	$S2 = 2 \times S1$	$S3 = 5 \times S1$
		Geometry $n^{\circ}1$ ( $S1 = 5.6 \text{ dm}^2$ )	Geometry $n^{\circ}2$ ( $S2 = 11.2 \text{ dm}^2$ )	Geometry $n^{\circ}3$ ( $S3 = 28 \text{ dm}^2$ )
Q1	Vapour Flow n°1 (Q1 = 6.5 gr/min)	Case n°1	Case n°4	Case n°7
$\mathbf{Q2} = 2 \times \mathbf{Q1}$	Vapour Flow n°2 ( Q2 = 13 gr/min )	Case n°2	Case n°5	Case n°8
Q3 ≈ 5 x Q1	Vapour Flow n°3 ( Q3 = 30 gr/min )	Case n°3	Case n°6	Case n°9

#### And for Every Case:

- ⇒ Same Tank Internal Volume (29 Liters) and Liquid Quantity (20 Liters)
- ⇒ Fixed Wall Temperature (Twall) fixed by "Empiric" Determination

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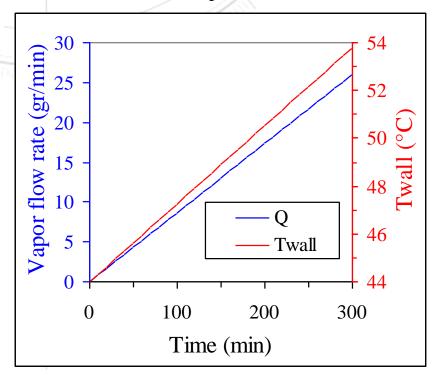


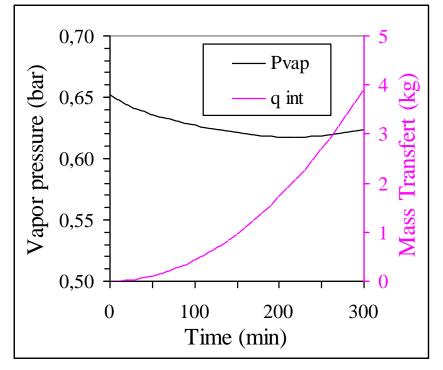
## RIModel - Exemple of Results for Case n°5

#### 1 - Boundary Conditions

- $\Rightarrow$  Collected Vapor Flow: Q2
- $\Rightarrow$  Evaporation Surface :  $S2 = 11.2 \text{ dm}^2$
- $\Rightarrow$  Fixed Wall Temperature : Twall

- 2 Simulation Results ( start from 44°C)
- ⇒ Vapor Pressure : Pvap
- $\Rightarrow$  Interfacial Mass Transfert : q int

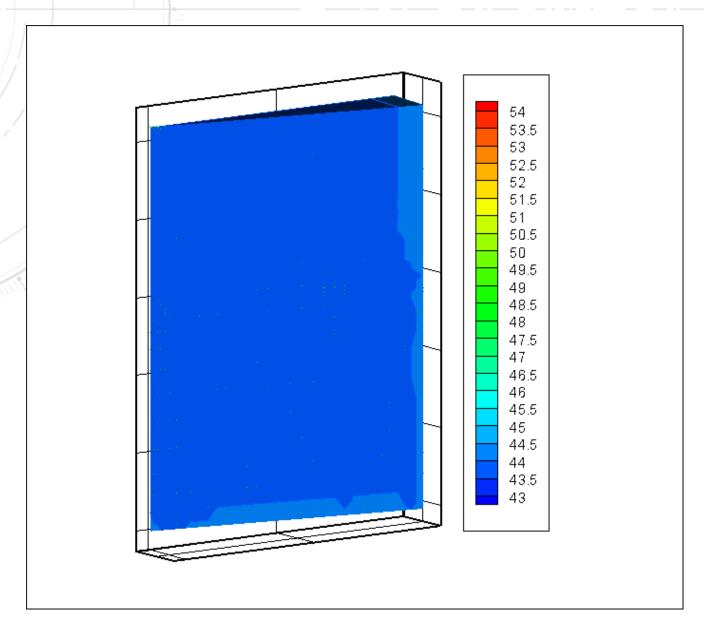




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qualiflow

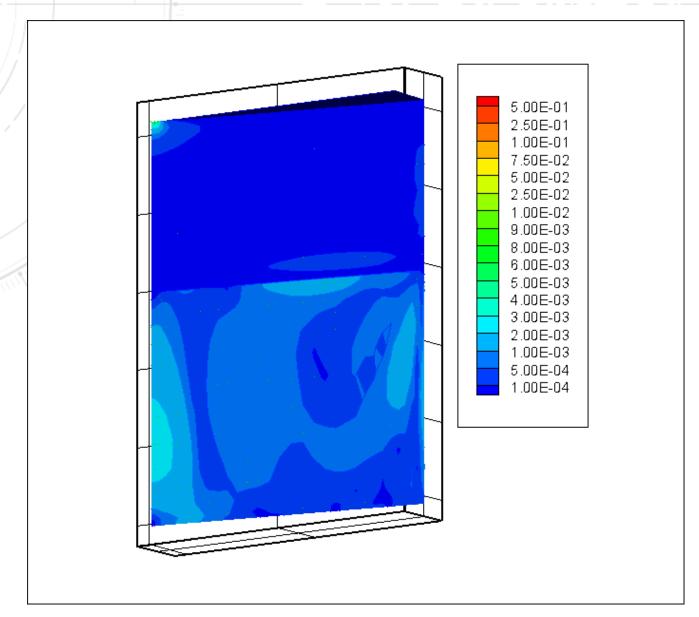
**Temperature** 



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qualiflow

<u>Velocity</u>



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## RIModel VALIDITY LIMIT - Subcooled Pool Boiling

- 1 The RIModel ⇒ Assumes Pool Subcooled Boiling Situation
- 2 The Transition Pool Subcooled Boiling to Nucleate Boiling
  - ⇒ Depend on Gap between Wall and Saturation Temperatures

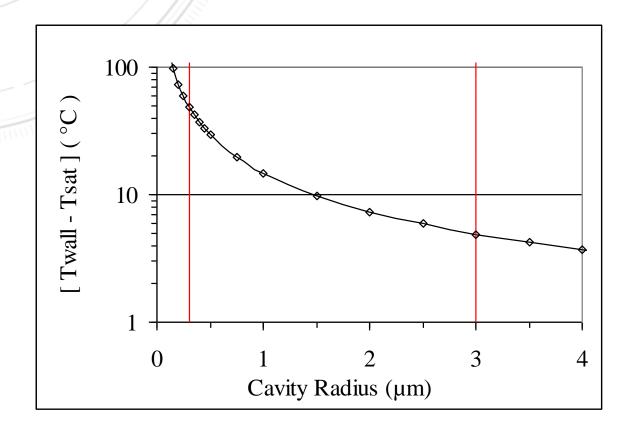
$$\Delta T = Twall - Tsat = [2 \cdot \sigma \cdot Tsat] / [\rho vap \cdot \Delta H vap \cdot R]$$

- ⇒ Depend on Nucleation Cavity Radius
- 3 Typical Cavity Radius in the µm Range
  - $\Rightarrow 0.3 \ \mu m < Cavity Radius : R < 3 \ \mu m \Rightarrow 50 \ ^{\circ}C > \Delta T > 5 \ ^{\circ}C$
  - ⇒ CONSEQUENCE: RIModel valid for Temperature Gap of about 20°C (and even more: no major disturbance induced by Beginning of Nucleate Boiling)



## RIModel VALIDITY LIMIT - Subcooled Pool Boiling

 $\Rightarrow$  Onset of Heterogeneous Nucleation  $\Rightarrow \Delta T = f(Cavity Radius)$ 

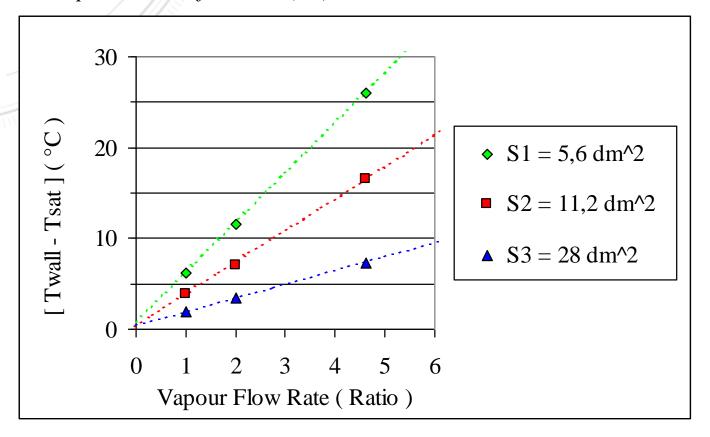


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## RIModel - Parametric Study Results - Temperature Gap

 $\Rightarrow$  Temperature Gap Dependence with Vapor Flow Rate ( Q ) and Evaporation Surface Size ( S )



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#### CONCLUSION

- 1 The RIModel for Evaporation of Liquids in Closed Tanks is:
  - ⇒ Based upon Heat Conductivity at Liquid / Vapor Interface by analogy with Melting or Solidification Process
  - ⇒ Assuming Pool Subcooled Boiling Situation
  - ⇒ Assuming to preserve Plane Interface during Evaporation
- 2 The Parametric Study \Rightarrow Quantify influence of Boundary conditions (Q, S, Twall)
  - $\Rightarrow$  For same couple (Q, Twall)  $\Rightarrow$  The larger (S)  $\Rightarrow$  The bigger (Pvap)
  - $\Rightarrow$  For same couple (Q, Pvap)  $\Rightarrow$  The larger (S)  $\Rightarrow$  The smaller (Twall)
- 3 Critical discussion of Numerical Results
  - $\Rightarrow$  With features of Boiling Process and Mechanism of Bubbles Generation  $\Rightarrow$  RIModel Limit of Validity